# MMEMORIAL UNIVERSITY <br> <br> PROC 5071: <br> <br> PROC 5071: <br> Process Equipment Design I 

Size and Shape of Particles

Salim Ahmed

## 1 Properties of solids particles

1.1 Why properties of solid particles are important?

- Materials are either solids or fluids.
- Process industries encounter handling of both solids and fluids.
- In process industries, the most relevant of solids are the small particles.
- An understanding of the characteristics of masses| of particulate solid in designing process and equipment is necessary for dealing with streams containing such solids


### 1.2 Characterization of solids

- Transportation and flow properties are involved in unit operations
- In terms of flow properties, fluids are often characterized based on
- density and
- viscosity
- What distinguishes fluids from solids are their shapelessness
- Solids are often characterized in terms of their - shape and
o size
- However, other properties are also important
- Density - fluidization
- Hardness, toughness, stickiness - crushing and grinding
- Thermal conductivity - heat transfer


## 2 Size of particles

### 2.1 How to define size?

- We often define size in terms of big and small
- What is big for a processing facility is too small for a mining operation
- What is small for a process facility is too big for a nano-operation


### 2.2 What does size mean?



Figure 1: Magnitude of particle size in gas-solid systems [Source: Solo (1990).

- What do the above sizes mean?
- If we say diameter, then typically it refers to spherical particles
- One can also mention cubic particles with side of certain length.


### 2.3 Characteristic dimensions of regular shaped particles

- For regular shaped particles, say spheres or cubes,I if one characteristic dimension is known, other properties say, surface area can be calculated
- For example, for a spherical particle with diameter $D$, the volume is $\frac{\pi}{6} D^{3}$ and surface area is $\pi D^{2}$.
- However, in real life, most particles are NOT of regular shape
- The Earth itself is not a perfect sphere; grains, fruits all are irregular in shape; and the human body itself is a vivid example of irregualr shape object.
- How to define size of irregular shape particles.
2.4 Characteristic dimensions of irregular shaped particles
- The fundamental problem is how to define a three dimensional irregular shaped particle by one dimension.
- In practice, it's not possible. However, by defining a size and a property defining its shape, a particle can be characterized.
- First the shape needs to be defined.


## 3 Shape of particles

3.1 Uniqueness of the sphere

- Symmetry is a unique feature of the sphere
- A sphere, despite being three dimensional, can be characterized by just one number, its diameter.
- Also, the sphere has the smallest surface area to volume ratio. Surface area is one of the important particle property for chemical and process industries.
- The sphere is taken as standard.
- Particle shapes are compared with the sphere by defining their sphericity.


### 3.2 Sphericity

- The sphericity is defined as

$$
\begin{equation*}
\Phi_{s}=\frac{s_{s p h}}{s_{p}} \tag{1}
\end{equation*}
$$

$s_{s p h}$ : surface area of sphere of same volume as particle
$s_{p}$ : surface area of particle

- Defining $v_{p}$ as volume of one particle and $v_{s p h}$ as volume of the sphere, where $v_{s p h}=v_{p}$

$$
\begin{equation*}
\Phi_{s}=\frac{s_{s p h} / v_{s p h}}{s_{p} / v_{p}} \tag{2}
\end{equation*}
$$

- For a sphere, $s_{s p h}=\pi D_{s p h}^{2}$ and $v_{s p h}=\frac{\pi}{6} D_{s p h}^{3}$ | where $D_{s p h}$ is the diameter of the sphere.
- So we have $\frac{s_{s p h}}{v_{s p h}}=\frac{6}{D_{s p h}}$ giving

$$
\begin{equation*}
\Phi_{s}=\frac{6 / D_{s p h}}{s_{p} / v_{p}} \tag{3}
\end{equation*}
$$

- For irregular shaped particles, an equivalent diameter, $D_{p}$, is defined as the diameter of a sphere with the same volume as that of the particle. So we have $D_{s p h}=D_{p}$ which gives

$$
\begin{equation*}
\Phi_{s}=\frac{6 / D_{p}}{s_{p} / v_{p}} \tag{4}
\end{equation*}
$$

- Note that sphericity of particles of a particular shape is independent of their sizes
- sphericity of all cubes are the same


### 3.3 Notes on equivalent/nominal diameter

- An equivalent diameter is defined as the diameter of a sphere of equal volume.
- However, sometimes equivalent diameter of cer-I tain shapes are defined arbitrarily; e.g. equivalent diameter of a cube is defined as the length of one side.
- Sometimes the second-largest dimension is taken as equivalent diameter; e.g. equivalent diameter of cylinders are taken as their diameter.
- For fine granular materials, equivalent diameter is determined based on screen size.


### 3.4 Workbook: Sphericity of particles

Determine the sphericity of

1. a sphere with diameter of 2 cm .
2. a cube with one side $D_{p} \mathrm{~cm}$.
3. a cylinder with diameter of $D_{p} \mathrm{~cm}$ and $L=D_{p}$.

### 3.5 Workbook: Sphericity of hemispheres

> You have a sphere with diameter $D$. if you cut the sphere into two halves, determine the sphericity of each of the hemispheres.
3.6 What's the use of sphericity

- Question arises that if we need surface area and volume to estimate the sphericity, then what's its use?
- In reality, the sphericity is used to calculate other properties, e.g. surface area.
- Sphericity is independent of size of particles having the same shape.
- If you know or determine the sphericity of one particle, you can use it for bulk particles with different sizes but the same shape.
- Many naturally existing particles, e.g. minerals, may have different sizes but have the same shape.
- Crystals may grow to various sizes while having the same shape.

Table 1: Sphericity of miscellaneous materials (Source: Perry's Chemical Engineering Handbook).
Materials
Spheres, cubes, short
cylinders ( $L=D_{p}$ )
Raschig rings $\left(L=D_{p}\right)$
$L=D_{o}, D_{i}=0.5 D_{o}$
0.58
$L=D_{o}, D_{i}=0.75 D_{o} \quad 0.33$
Berl saddles 0.3
Ottawa sand $\quad 0.95$
Rounded sand 0.83
Coal dust 0.73
Flint sand $\quad 0.65$
Crushed glass $\quad 0.65$
Mica flakes $\quad 0.28$

### 3.7 Volume shape factor

- Volume of any particle is proportional to the its diameter cubed

$$
\begin{equation*}
v_{p} \propto D_{p}^{3} \tag{5}
\end{equation*}
$$

- For a given shape we get

$$
\begin{equation*}
v_{p}=a D_{p}^{3} \tag{6}
\end{equation*}
$$

- Here $a$ is called the volume shape factor. For a sphere $a=\frac{\pi}{6}=0.5236$, for a cube $a=1$.
- The volume shape factor is independent of size; it depends on the shape.


## 4 Properties of bulk particles

- We determine the size and shape of particles to calculate other particle characteristics, e.g. surface area, or properties of a mass of particles.
- For irregular shape particles, it is not straightforward to determine surface area of particles.
- For many particles, we can determine the size by screen analysis.
- Also for certain particles, the sphericity is known.I
- Now, how to determine the total surface area of a mass of particles with known particle density?
4.1 Surface area of a bulk of equal size irregular shape particles
- Total surface area of a mass of particles

$$
\begin{equation*}
A=N \times s_{p} \tag{7}
\end{equation*}
$$

where, $N$ is the number of particles and $s_{p}$ is the surface area of one particle.

- The number of particles can be obtained by dividing the total mass by the mass of one particle

$$
\begin{equation*}
N=\frac{m}{\rho_{p} v_{p}} \tag{8}
\end{equation*}
$$

- So we have

$$
\begin{equation*}
A=\frac{m}{\rho_{p} v_{p}} \times s_{p} \tag{9}
\end{equation*}
$$

- However, it is not straightforward to measure the surface area or volume of individual particles because of their irregular shape.
- If the sphericity, $\Phi_{s}$ and the diameter of particles, $D_{p}$ are known, we can relate $s_{p}$ and $v_{p}$ to $\Phi_{s}$

$$
\begin{aligned}
\Phi_{s} & =\frac{6 / D_{p}}{s_{p} / v_{p}} \\
\frac{s_{p}}{v_{p}} & =\frac{6}{\Phi_{s} D_{p}}
\end{aligned}
$$

- This gives

$$
\begin{align*}
A & =\frac{m}{\rho_{p}} \times \frac{s_{p}}{v_{p}} \\
& =\frac{6 m}{\Phi_{s} \rho_{p} D_{p}} \tag{11}
\end{align*}
$$

4.2 Properties of a bulk of particles with different sizes

- A bulk containing particles with different sizes, the size distribution is normally expressed as a histogram as shown below.
- The histogram represents mass fraction of particles of different sizes.
- An alternate representation depicting cumulative mass fraction smaller than stated sizes is also used.
4.3 Surface area of a bulk of particles with different sizes
- The total surface area of a mass of particles can be represented as a sum of the surface areas of masses of different sizes.
- For a size $\bar{D}_{p_{i}}$, if the mass fraction of particles is $x_{i}$, then the surface area of all particles of this size is

$$
\begin{equation*}
A_{i}=\frac{6 m_{i}}{\Phi_{s} \rho_{p} \bar{D}_{p_{i}}} \tag{12}
\end{equation*}
$$

- If there are $n$ number of sizes, total area is

$$
\begin{equation*}
A=\sum_{i=1}^{n} \frac{6 m_{i}}{\Phi_{s} \rho_{p} \bar{D}_{p_{i}}} \tag{13}
\end{equation*}
$$

- The specific surface area i.e. the surface area per unit mass of particles is given by

$$
\begin{equation*}
A_{w}=\frac{\frac{6}{\Phi_{s} \rho_{p}} \sum_{i=1}^{n} \frac{m_{i}}{\overline{D_{p_{i}}}}}{\sum_{i=1}^{n} m_{i}} \tag{14}
\end{equation*}
$$

- If the total mass is $m$, then $m_{i}=m x_{i}$ and we get

$$
\begin{equation*}
A_{w}=\frac{\frac{6}{\Phi_{s} \rho_{p}} \sum_{i=1}^{n} \frac{x_{i}}{\overline{D_{p_{i}}}}}{\sum_{i=1}^{n} x_{i}} \tag{15}
\end{equation*}
$$

- Note that if all of the screens are considered then $\sum_{i=1}^{n} x_{i}=1$ and the equation simplifies
to

$$
\begin{equation*}
A_{w}=\frac{6}{\Phi_{s} \rho_{p}} \sum_{i=1}^{n} \frac{x_{i}}{\bar{D}_{p_{i}}} \tag{16}
\end{equation*}
$$

5 Average size of bulk particles

There are a number of ways to define the average particle size.

1. Surface mean diameter
2. Arithmetic mean diameter
3. Mass mean diameter
4. Volume mean diameter

### 5.1 Surface mean diameter

- defined as the diameter of the same number of equal size particles that would give the same surface area as that of the mass.
- If there were only one size particles with diam-
eter $D_{s}$, the specific surface area would be

$$
\begin{equation*}
A_{w}=\frac{6}{\Phi_{s} \rho_{p} \bar{D}_{s}} \tag{17}
\end{equation*}
$$

- The specific surface of the mass is given by

$$
\begin{equation*}
A_{w}=\frac{\frac{6}{\Phi_{s} \rho_{p}} \sum_{i=1}^{n} \frac{x_{i}}{\overline{D_{p_{i}}}}}{\sum_{i=1}^{n} x_{i}} \tag{18}
\end{equation*}
$$

- Equating these two

$$
\begin{equation*}
\bar{D}_{s}=\frac{\sum_{i=1}^{n} x_{i}}{\sum_{i=1}^{n} \frac{x_{i}}{\bar{D}_{p_{i}}}} \tag{19}
\end{equation*}
$$

- If all of the screens are considered, then

$$
\begin{equation*}
\bar{D}_{s}=\frac{1}{\sum_{i=1}^{n} \frac{x_{i}}{\bar{D}_{p_{i}}}} \tag{20}
\end{equation*}
$$

### 5.2 Arithmetic mean diameter

- defined as the weighted average of the diameters where the number of particles is used as
the weighting factor.

$$
\begin{equation*}
\bar{D}_{N}=\frac{\sum_{i=1}^{n} N_{i} D_{p_{i}}}{\sum_{i=1}^{n} N_{i}} \tag{21}
\end{equation*}
$$

### 5.3 Mass mean diameter

- defined as the weighted average of the diameters where the mass fraction of particles belonging to the stated size is used as the weighting factor.

$$
\begin{equation*}
\bar{D}_{w}=\frac{\sum_{i=1}^{n} x_{i} D_{p_{i}}}{\sum_{i=1}^{n} x_{i}} \tag{22}
\end{equation*}
$$

5.4 Volume mean diameter

- defined as the diameter of the same number of equal size particles that would give the same volume as that of the mass.
- If there were only one size particles with diameter $\bar{D}_{v}$, the volume would be

$$
\begin{equation*}
v=N \times a \bar{D}_{v}^{3} \tag{23}
\end{equation*}
$$

- The volume of the mass is given by

$$
\begin{equation*}
v=\frac{\sum_{i=1}^{n} m_{i}}{\rho_{p}} \tag{24}
\end{equation*}
$$

- Equating these two

$$
\begin{equation*}
N \times a \bar{D}_{v}^{3}=\frac{\sum_{i=1}^{n} m_{i}}{\rho_{p}} \tag{25}
\end{equation*}
$$

- $N$ is related to the known parameters as

$$
\begin{equation*}
N=\sum_{i=1}^{n} \frac{m_{i}}{\rho_{p} a \bar{D}_{p_{i}}^{3}} \tag{26}
\end{equation*}
$$

- Using these relations and $x_{i}=m_{i} / m$ we get

$$
\begin{equation*}
D_{v}=\left[\frac{\sum_{i=1}^{n} x_{i}}{\sum_{i=1}^{n} \frac{x_{i}}{\bar{D}_{p_{i}}^{3}}}\right]^{\frac{1}{3}} \tag{27}
\end{equation*}
$$

## 6 Points to ponder

- For an irregular shape particle, what does 'size' mean?
-Why is 'shape' important for particles?
- Why do we consider 'sphericity' and what is its use?
- Why do we need to define mean diameter in different ways?


## References

1. W. L. McCabe, J. C. Smith, P. Harriott. (2005).I Unit Operations of Chemical Engineering, 7th Edition, McGraw Hill, New York, USA. ISBN13: 978-0-07-284823-6
2. D. Green and R.H. Perry. 2007. Perry's Chemical Engineers' Handbook, 8th Edition, McGraw-I Hill. ISBN-13: 9780071422949
3. J. F. Richardson, J. H. Harker, J. R. Backhurts (2002) Coulson and Richardson's Chemical Engineering - Particle Technology and Separation Processes, Vol. 2, Fifth Ed., Butterworth-Heinemann Oxford, UK. ISBN 0750644451
